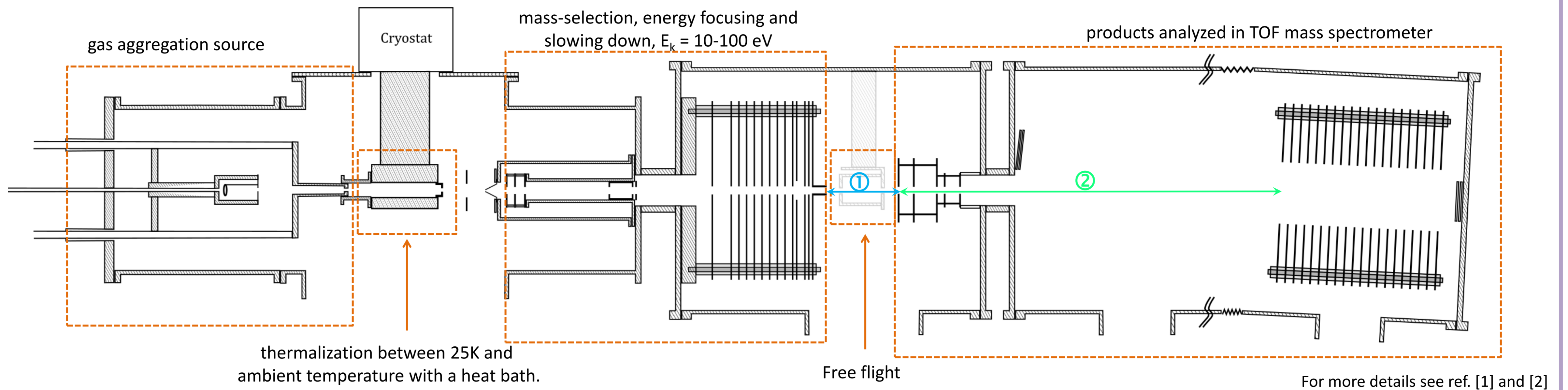


Experimental setup

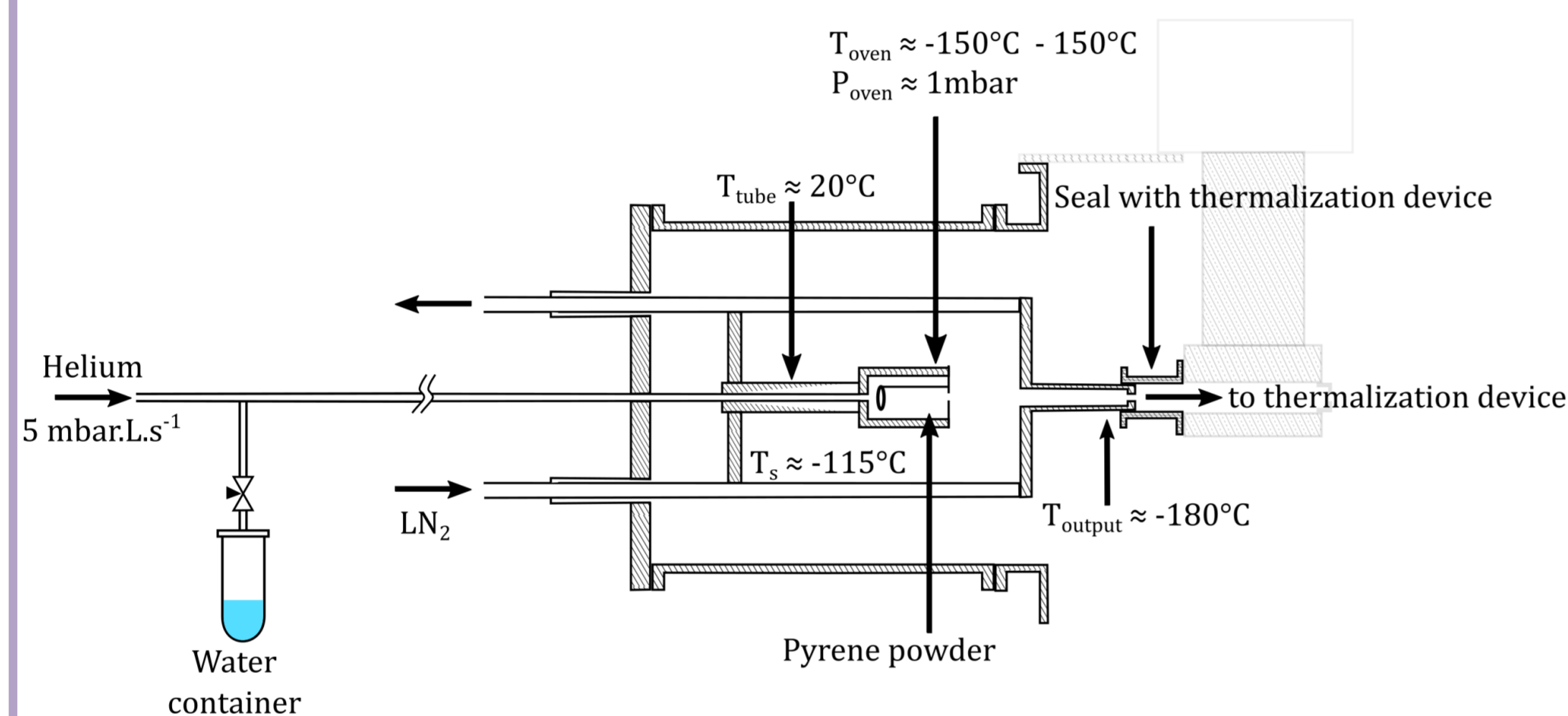


Clusters production

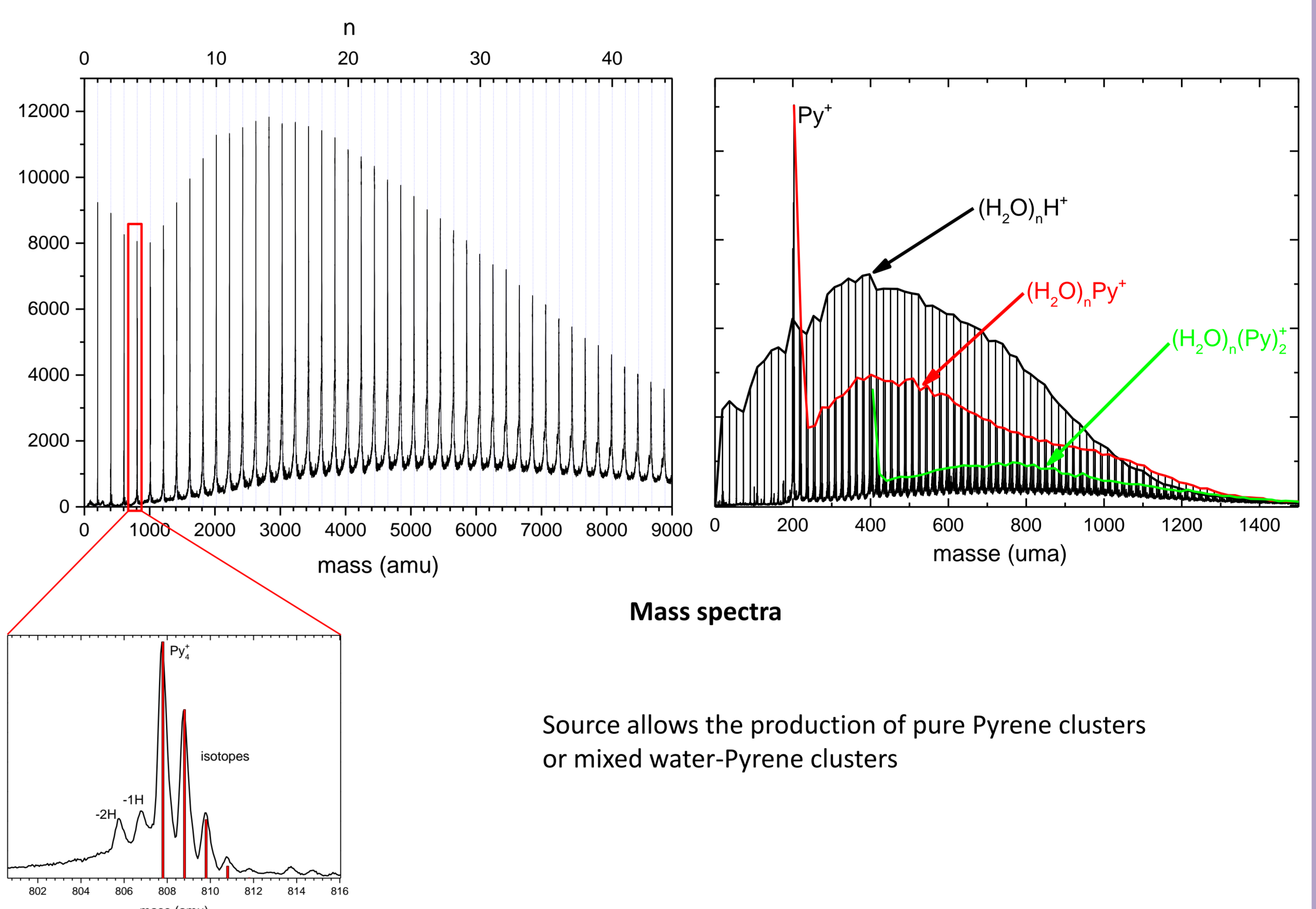
Source conditions

Clusters are produced in a gas aggregation source. Ionization by either :

- a discharge electrode, $I \approx 100\mu\text{A}$, $V \approx 1$ kV
- an electron gun, $I = 2.5$ A, $V = -100$ V



Observation of dehydrogenation



Evaporation rates of pure Pyrene clusters

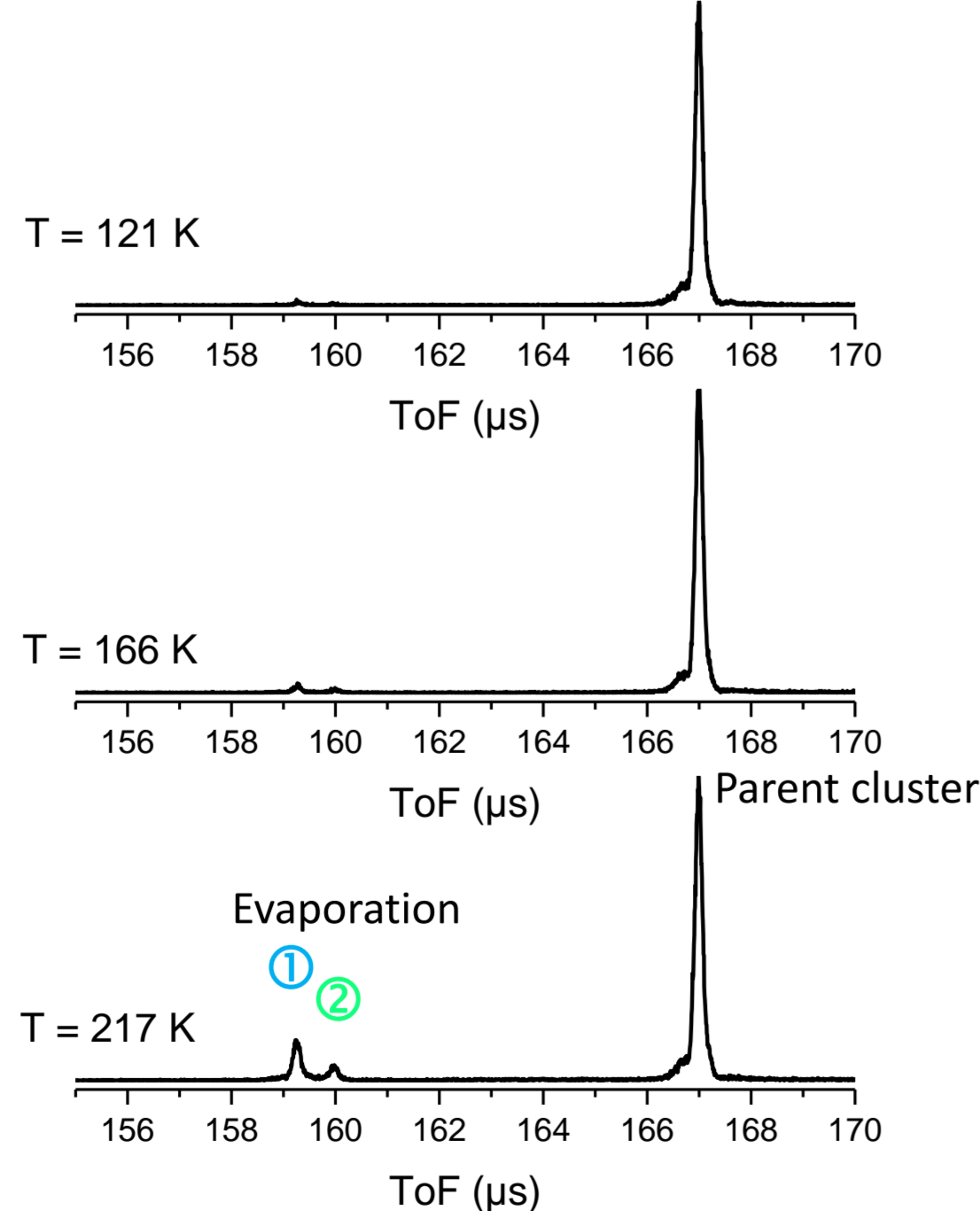
Principle of measurement

Temperature controlled $(\text{Py})_n^+$ clusters are mass-selected and slowed down. Mass-spectra show the appearance of evaporation products as the cluster temperature is increased.

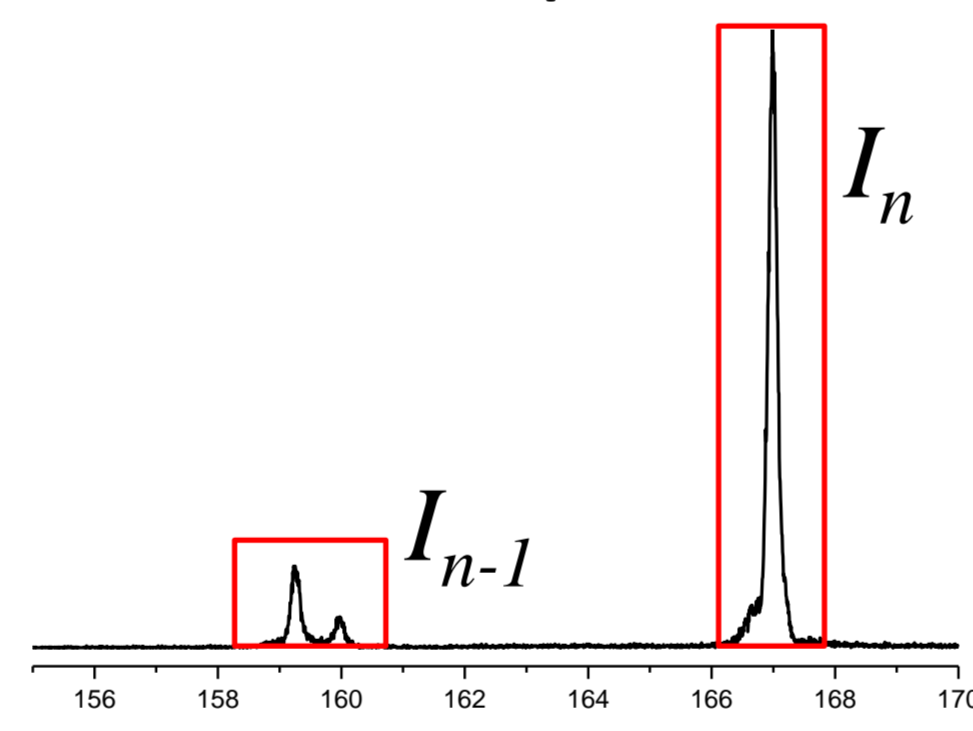
Evaporations are detected if they take place between the end of the slowing down and the entrance of the reflectron (①+②).

The $n-1$ peak has a double structure due to evaporations taking place after the second acceleration and before the reflectron (②).

Example: $(\text{Py})_{11}^+$ @ 22 eV



Extraction of the evaporation rate $W(T)$



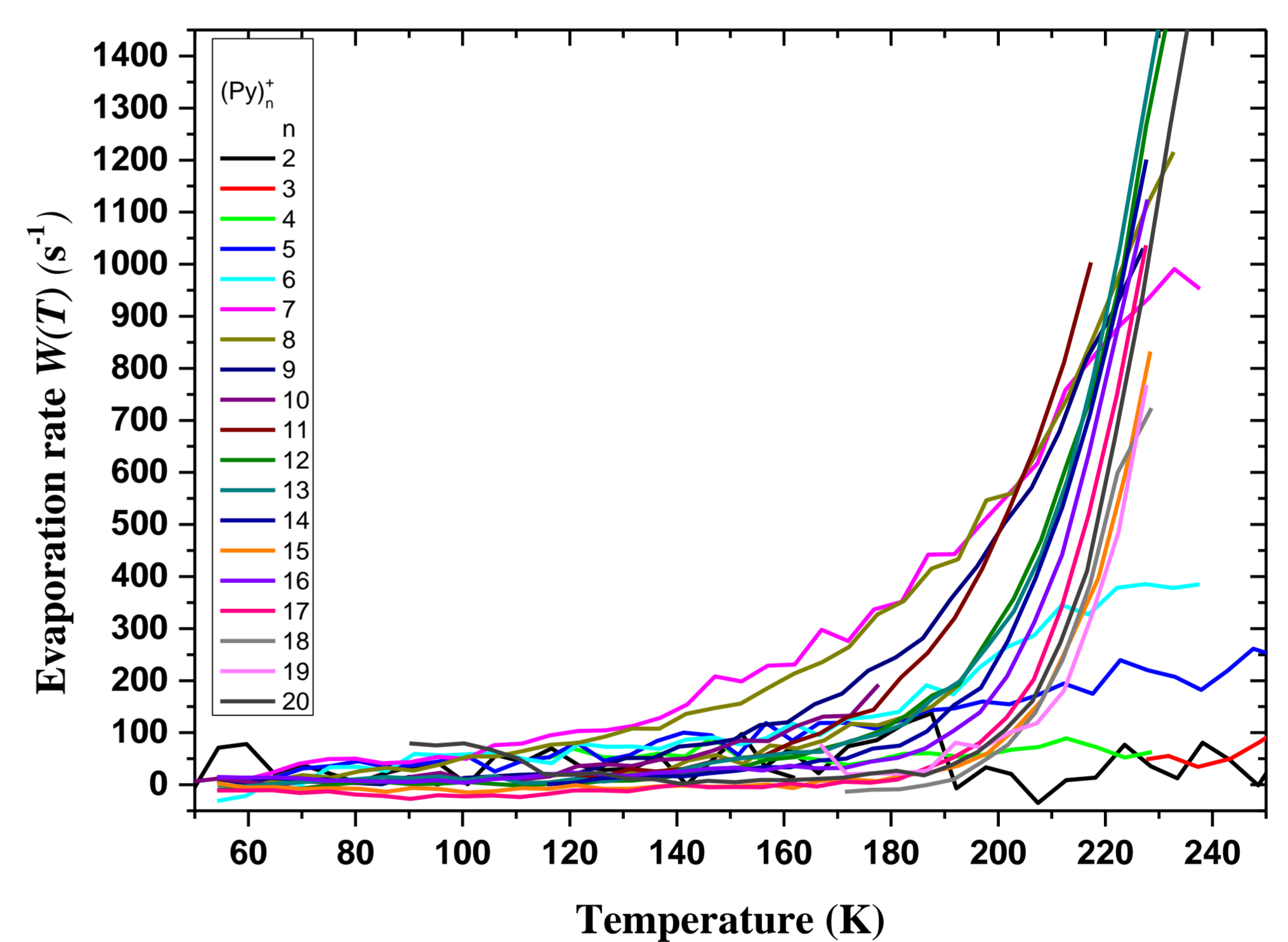
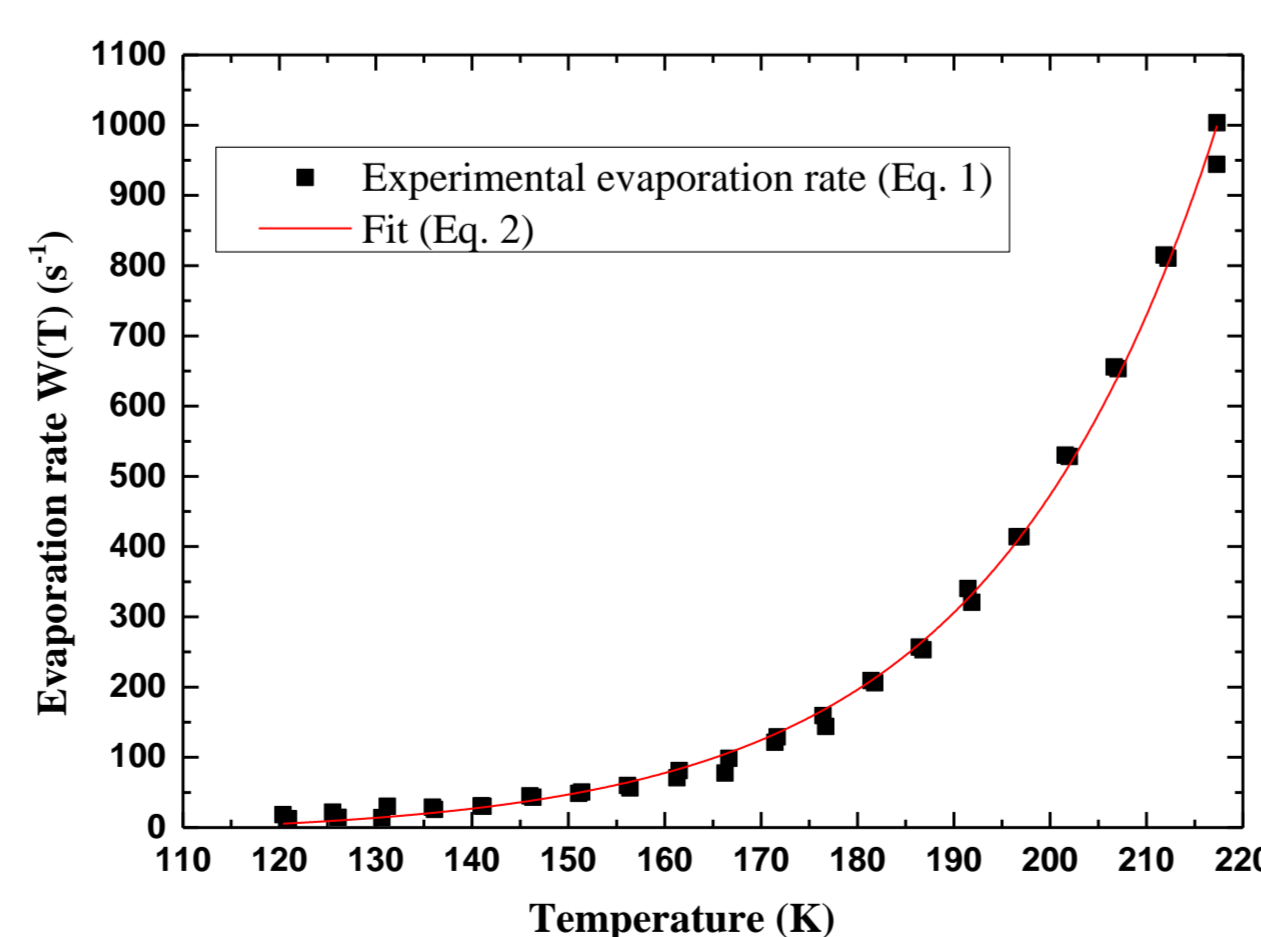
$$I_n = (I_n + I_{n-1})e^{-W(T)t}$$

$$\Rightarrow W(T) = -\ln\left(\frac{I_n}{I_n + I_{n-1}}\right) \times \frac{1}{t} \quad (\text{Eq. 1})$$

t is the time of flight in the region ①+②. For $n=11$ pyrene clusters @ 22eV, $t = 160 \mu\text{s}$.

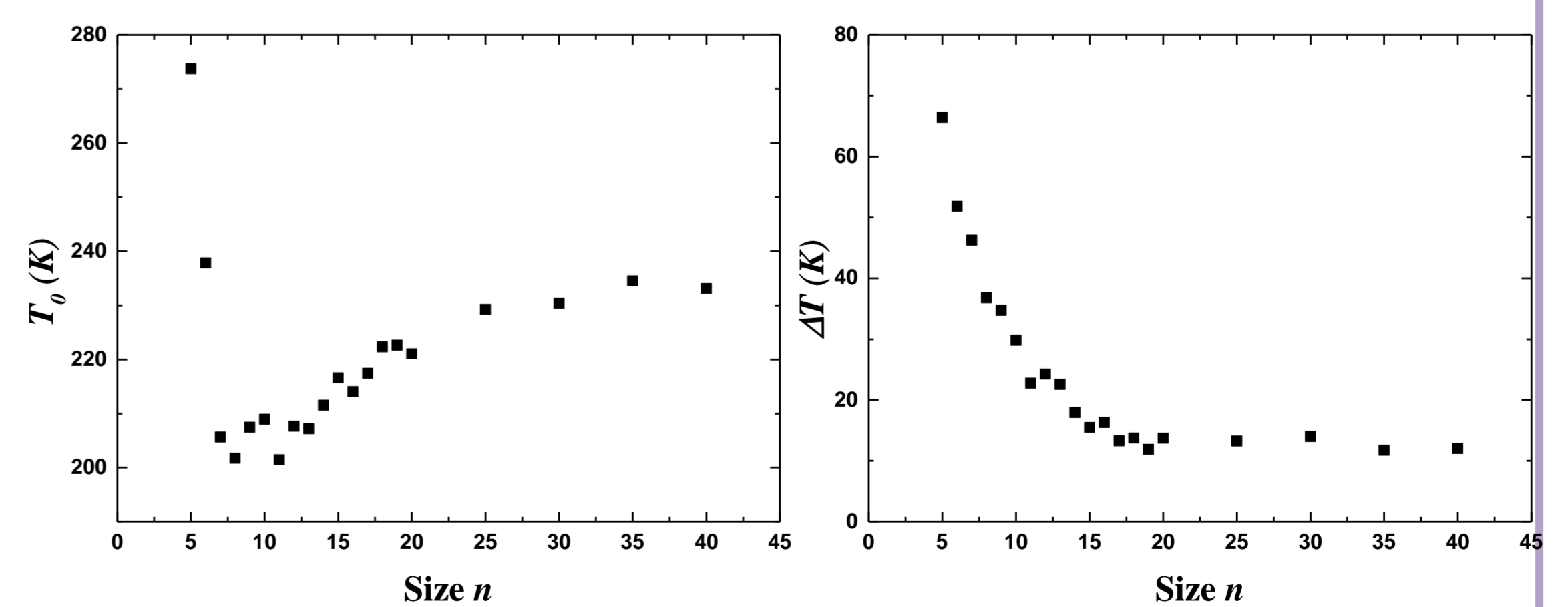
Experimental data are fitted by the empirical function:

$$f(T) = W_0 + 500e^{(T-T_0)/\Delta T} \quad (\text{Eq. 2})$$



Above: evaporation rates for all sizes as a function of temperature.

Below: these curves are fitted with Eq. 2 to extract the threshold temperature T_0 and the width ΔT :



For $n=2-4$, no evaporation observed up to 273K. Above $n=6$, the threshold temperature T_0 increases with size. Around $n=20$, evaporation rates stop evolving with size. $W(T)$ get steeper as the size increases.

What's next ?

- Phase space theory and/or RRKM calculations are needed to interpret our results
- Collision induced dissociation of pyrene clusters could also give threshold energies for dissociation
- Study of mixed water-pyrene clusters

References

- [1] F. Chirot, S. Zamith, P. Labastie, and J.-M. L'Hermite, Rev. Sci. Instrum. **77**, 063108 (2006)
- [2] I. Braud, S. Zamith, and J.-M. L'Hermite, Rev. Sci. Instrum. **88**, 043102 (2017)

